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have a positive charge; located in the  
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Composition 1. 100 washers 0.110 g 1 washer = 11.0 g (assuming 100 washers is exact) 100. g 1 washer 0.110 g = 909 washers 2. 500. g 1 cork 1.63 g = 306.7 = 307 corks 500. g 1 stopper 4.31 g = 116 stoppers 1 kg

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(1000 g) of corks contains (1000 g  
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is the condition that results from changes in the independent variable.

\_\_\_\_\_ results in an increase in the amount of living material and the formation of new structures. 3 Take 5

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6. (c) 7. (d) 8. (b) 9. (a) 10. (c) NCERT  
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1. Why standard entropy of an  
elementary substance is not zero  
whereas the standard enthalpy of  
formation is taken as zero? Answer: A  
substance has a perfectly ordered

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Arrangement only at absolute zero.  
Hence, entropy is zero only at ...

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Thermodynamics Thermodynamics is  
the part of Chemistry which deals with  
exchange of energy in the form of heat

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and temperature. William Thomson coined this term in 1794 by combining two Greek words Thermes which is heat and Dynamikos meaning powerful.

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## Chapter 6 Equilibrium Chemistry 211

the reaction we monitor the mass of  $\text{Ca}^{2+}$  remaining in solution and the mass of  $\text{CaCO}_3$  that precipitates, the result looks something like Figure 6.1. At the start of the reaction the mass of  $\text{Ca}^{2+}$  decreases and the mass of  $\text{CaCO}_3$  increases. Eventually the

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Reaction reaches a point after which  
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