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2 (g) + 2H₂O () CH₄ (g) + 2O₂ (g) During the reaction the

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temperature of the reaction mixture decreases. 6.1 Define the term enthalpy change. (2) 6.2 Does the enthalpy change (ΔH) for this reaction have a positive or negative value? Explain the answer by referring to the energy involved. (2)

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